## CHEMISTRY OF SULFONYIMETHYLISOCYANIDES. SIMPLE SYNTHETIC APPROACHES TO A NEW VERSATILE CHEMICAL BUILDING BLOCK.

 A.M. van Leusen, G.J.M. Boerma, R.B. Helmholdt, H. Siderius and J. Strating Department of Organic Chemistry, The University, Zernikelaan, Groningen, The Netherlands.
(Received in UK 17 April 1972; accepted for publication 3 May 1972)

Synthetic applications of isocyanides have become widespread in recent years.<sup>1</sup> In this communication we wish to report on the synthesis of tosylmethylisocyanide (TosMIC) (Ia) and related compounds. The succeeding papers<sup>2</sup> deal with the unique synthetic applications of Ia to the formation of a number of heterocyclic systems that are otherwise more difficult to prepare.

The first representative of this new class of isocyanides, tosylmethylisocyanide (Ia), was obtained some years ago (in 14% yield) when we irradiated tosyldiazomethane in liquid HCN.<sup>3</sup> In order to scrutinize the properties of I, a more profitable and less hazardous synthesis was much needed. Two such syntheses are now available: (i) the dehydration of sulfonyl-substituted secondary formamides (II, route a), and (ii) the sulfonylation of  $\alpha$ -lithicisocyanides (route b).

$$\begin{array}{c} \text{POCl}_{3} \text{ a} \end{pmatrix} \xrightarrow{\text{POCl}_{3} \text{ a}} \text{RSO}_{2}\text{CR'R"-N=C} \xrightarrow{\text{b}} \text{RSO}_{2}\text{F} + \text{R'R"C-N=C} \\ (\text{II}) & \text{glyme, Et}_{3}\text{N, } \text{O-10}^{\circ} & (\text{I}) & \text{-LiF} & \text{i} \\ \text{R"} = \text{H, R en R' see Table.} \end{array}$$

Compounds II are prepared conveniently by oxidation of the corresponding sulfides<sup>4</sup> (III), or, as has been shown recently<sup>5</sup>, by a Mannich condensation under forcing reaction conditions.

RSCR'R"NHCHO 
$$\xrightarrow{\text{MCPBA, 0}^{\circ}}$$
 II  $\leftarrow$  RSO<sub>2</sub>H + R'R"CO + H<sub>2</sub>NCHO  
(III) R" = H

The dehydration of II is readily achieved with FOCl<sub>3</sub> in dimethoxyethane -Et<sub>3</sub>N mixture. Compound Ic was obtained, quite recently, by Böhme and Fuchs<sup>6</sup> under somewhat different reaction conditions. Sulfonyl fluorides have been used successfully for the sulfonylation of certain carbon nucleophiles.<sup>7</sup> This observation, combined with the recently established nucleophilic behaviour of  $\alpha$ -lithioisocyanides<sup>8</sup>, led to the alternate synthesis of I <u>via</u> route b (see Table, compounds Ia and If). Route a, however, offers the advantage of avoiding the use of evil-smelling volatile isocyanides.

	R	R'	R"	m.p.* ℃	route a %	route b %	alkyl. %	N=C cm <sup>-1</sup>	δH <sub>α</sub> ppm	
Ia	p-tolyl	н	Н	116-7 <sup>0</sup>	80	87		2150	4.59	
Ib	benzyl	н	н	103-6°	60			2150	4.50	
Ie	phenyl (re	ef.6)H	H	88 <sup>0</sup>	66			2150		
Id	n-butyl	Н	н	46-47 <sup>0</sup>	80			<b>21</b> 55	4.60	
Ie	t-butyl	Н	Н	104 <b>-</b> 5 <sup>0</sup>	66*			2155	4.60	
If	p-tolyl	phenyl	Н	124-5 <sup>0</sup>		30		2140	5.60	
Ig	p-tolyl	benzyl	benzyl	112 <b>-</b> 3 <sup>0</sup>			80	2140		
Ih	p-tolyl	ethyl	ethyl	85 <b>-</b> 6°			80	2145		

\* Slight decomposition occurs at the m.p.: Id and Ie decompose slowly on standing at R.T. \* The dehydration reaction was carried out at -40°.

 $\alpha$ -Alkylation of TosMIC leads to further  $\alpha$ -isocyanosulfones. Doubly alkylated products (Ig,h) are obtained in high yield, using two moles of both base (NaH in DMSO/Et<sub>2</sub>0) and alkylhalide (PhCH<sub>2</sub>Br and EtI, resp.). However, pure monoalkylated derivatives are difficult to obtain (with equivalent quantities of reactants). These results parallel similar observations by Schöllkopf, <u>et.al</u>., with ethyl isocyanoacetate.<sup>9</sup>

## REFERENCES

1. A recent review: Isonitrile Chemistry, I. Ugi ed., Academic Press, New York 1971.

2. This journal, two succeeding papers.

3. A.M. van Leusen and J. Strating, Quart. Rep. Sulfur. Chem., 5, 67 (1970).

- Sulfides III, new compounds with exception of IIIa, were prepared according to B.D. Vineyard, J. Chem. Eng. Data, 11, 620 (1966).
- 5. T. Olijnsma, J.B.F.N. Engberts and J. Strating, <u>Rec. Trav. Chim.</u>, <u>91</u>, 209 (1972).

6. H. Böhme and G. Fuchs, Chem. Ber., 103, 2775 (1970).

7. See A.M. van Leusen, B.A. Reith, A.J.W. Iedema and J. Strating, Rec. Trav. Chim., 91, 37 (1972).

8. U. Schöllkopf and P. Böhme, Angew. Chem., Int. Eng. Ed., 10, 491 (1971).

9. U. Schöllkopf, D. Hoppe and R. Jentsch, Angew. Chem., Int. Eng. Ed., 10, 331 (1971).